

FACULTY OF HEALTH AND APPLIED SCIENCES

DEPARTMENT OF HEALTH SCIENCES

QUALIFICATION: BACHELOR OF MEDICAL LABORATORY SCIENCES				
QUALIFICATION CODE: 08BMLS		LEVEL: 6		
COURSE CODE: CLC621S		COURSE NAME: CLINICAL CHEMISTRY 2B		
SESSION:	NOVEMBER 2019	PAPER:	THEORY	
DURATION:	3 HOURS	MARKS:	100	

FIRST OPPORTUNITY EXAMINATION QUESTION PAPER			
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MODERATOR:	MR MAURICE NYAMBUYA		

	INSTRUCTIONS	
1.	Answer ALL the questions.	
2.	Write clearly and neatly.	
3.	Number the answers clearly.	

PERMISSIBLE MATERIALS

1. NON PROGRAMMABLE CALCULATOR

THIS QUESTION PAPER CONSISTS OF FIVE PAGES (Including this front page)

SECTION A [30]

QUES	STION 1		[10]
1.0	Write short notes on the following:		
1.1	Faecal fat analysis		(5)
1.2	Special considerations when measuring serum ammonia.		(5)
QUES	STION 2		[10]
2.0	Review the following sets of results and	d answer the questions that follow:	
Patie	nt A	Patient B	
Total	Bilirubin: 230umol/L	Total Bilirubin: 45umol/L	
Direct	t Bilirubin: 1umol/L	Direct Bilirubin: 22umol/L	
Total	Protein: 63g/L	Total Protein: 55g/L	
Albun	nin: 42g/L	Albumin: 30g/L	
ALT: 5	5U/L	ALT: 800U/L	
AST: 8	38U/L	AST: 390U/L	
GGT:	40U/L	GGT: 133U/L	
ALP: 8	80U/L (Ref range:0-249U/L)	ALP: 66U/L (Ref range:40-130U/L)	
2.1	Indicate which patient is more likely to	suffer from haemolytic disease of the new	
	born. Motivate your answer.	,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	(2)
2.2	Which patient is more likely to produce pale stools? Motivate your answer.		(2)
2.3	,		
	each of patient A and B.		(2)
	a. Urine urobilinogen		
	b. Urine bilirubin	š	
2.4	Comment on the AST/ALT ratio of patie	nt A.	(2)
2.5	Assuming that your laboratory perform	s ionised Ca ⁺⁺ measurements only, would you	on (*)
	correct your calcium measurement? Motivate your answer.		

[10] **QUESTION 3** 3.0 Manual total protein assays were performed and the following absorbance readings were obtained: Patient A: Absorbance = 0.163 Patient B (1/3 dilution): Absorbance = 0.125 Control Absorbance = 0.291 Standard Absorbance = 0.235 Concentration = 75g/L 3.1 Calculate the concentration of total protein for Patient A and B. Round off your final (4)answers to 2 decimal places. 3.2 Which patient's result (A or B) is most likely to be observed in a case of: (2)Multiple myeloma a. Nephrotic syndrome 3.3 The package insert of the control states a mean of 72.5g/L and SD=6.25. Showing

SECTION B [30]

all calculations, motivate whether or not patient A and B's results can be accepted.

(4)

QUESTION 4 [30]

4.0 The following are laboratory results for an adult male patient from the clinical chemistry department. Review the results and answer the questions which follow.

		The state of the s
Serum/Plasma		
Sodium	141	136-146 mmol/L
Potassium	7.7	3.0-5.5 mmol/L
Chloride	107	90-110 mmol/L
CO ₂	22	20-30 mmol/L
Jrea	28.9	2.7-8.3 mmol/L
Creatinine	645	90-115 μmol/L
Glucose	19.8	3.5-5.5 mmol/L
Total protein	66	65-80 g/L
Albumin	28	35-50 g/L
Calcium	2.03	2.15-2.45 mmol/L
Osmolality	280	mOsm/kg
Serum Iron	11.2	11-17 μmol/L
Transferrin	3.5	2.0-3.6 g/L
Cholesterol	6.3	<5.2 mmol/L
Triglyceride (fasting)	3.53	<2.26 mmol/L
HDL	0.9	>1.2 mmol/L
24hr Urine		
Volume	1.3 L	
Creatinine	6	mmol/L

CONCENTRATION

REFERENCE RANGE

(2)

(2)

ANALYTE

4.3

4.4

Albumin/globulin ratio

Osmolal gap

4.5	Creatinine clearance	(3)
4.6	TIBC	(2)
4.7	% transferrin saturation	(2)
4.8	LDL	(3)
4.9	Comment on the patient's renal status	(2)
4.10	Briefly explain the causes of a low serum sodium	(5)
4.11	Discuss the causes of a raised serum potassium	(5)
	SECTION C [40]	
QUES	TION 5	[10]
5.0	Discuss the tests used for short, medium and long term monitoring of diabetes.	(10)
QUES	TION 6	[10]
6.0	Blood gas samples are very delicate and are to be treated as 'urgent samples'. Outline	
	the special conditions considered during arterial blood gas analysis.	(10)
QUES	TION 7	[10]
7.0	Give a detailed description of how the body degrades haemoglobin and the fate of	
	molecules produced during the catabolic process.	(10)
QUES	TION 8	[10]
8.0	Using relevant examples illustrate how enzymes are used as reagents in the clinical	
	chemistry laboratory.	(10)

END OF EXAMINATION

TOTAL 100 MARKS